THE STRUCTURE OF A NOVEL TERPENOID ACID "3ξ-HYDROXY-4ξ,9-DIMETHYL-6E,9E-DODECADIENEDIOIC ACID" ISOLATED FROM TURKISH TOBACCO.

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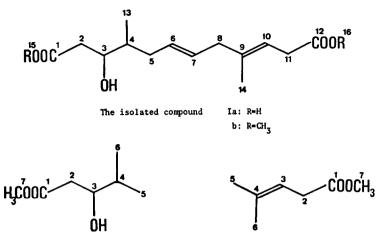
In the course of our studies on aroma constituents of Turkish tobacco,¹⁾ a new anomalous terpenoid acid " 3ξ -hydroxy- 4ξ ,9-dimethy1-6E,9E-dodecadienedioic acid" (Ia) was isolated. In this communication, we wish to report the isolation and structural elucidation of this new compound.

Repeated silicic acid column chromatagraphy of the acidic part of an isopropylether extract from cured Turkish tobacco leaves and purification by preparative gas chromatography after esterification with diazomethane, furnished the dimethyl ester of the new compound (Ib, $[\alpha]_n^{23}$ -7.9(c=2.0 in CH₃OH)).

The complete decoupling ¹³C-nmr showed the presence of fifteen lines. One of these lines (δ_{C} 51.6(q)) was revealed as a two carbon signal by the decoupling without NOE and resolved into two lines after adding the shift reagent "Eu(fod)₃", indicating that this compound had sixteen carbon atoms. Precise mass determination of the highest peak at m/e 280 in the double forcus mass spectrum (found 280.1679;calcd 280.1673), caused by a loss of water from the molecular ion, established that the molecular formula was $C_{16}H_{26}O_{5}$, which was also confirmed by the presence of a very weak peak at m/e 298 in single forcus mass spectrum.

The five oxygen atoms in the molecule, as judged from ir, 13 C- and 1 H-nmr,²) were attributable to two methyl esters (1740cm⁻¹, δ_{C} 173.5(s), 51.6(q), δ_{H} 3.74(s); 1735cm⁻¹, δ_{C} 172.6(s), 51.6(q), δ_{H} 3.69(s)) and a hydroxy group (3500cm⁻¹) attached to a methine (δ_{C} 71.2(d), δ_{H} 3.80(m)). Four unsaturations were due to two carbonyls in the methyl esters, a trans disubstituted double bond (972cm⁻¹, δ_{C} 129.4(d), 129.9(d), δ_{H} 5.44(m)) and a trisubstituted double bond (840cm⁻¹, δ_{C} 138.0(s), 116.0(d), δ_{H} 5.35(t)). Hence, the carbon skeleton of this compound was a chain structure and had no conjugated system because of the absence of the absorption at 210-360nm in uv spectrum and non shift of the carbonyl absorption in ir spectrum.

Of the two methyls revealed in ¹³C- and ¹H-nmr, one $(\delta_{C}^{14.9}(q), \delta_{H}^{0.88}(d))$ was attached to a methine carbon $(\delta_{C}^{38.2}(d), \delta_{H}^{1.64}(m), \text{ irradiation at 1.64 simplified the d. at 0.88 to a s.) and the other <math>(\delta_{C}^{16.1}(q), \delta_{H}^{1.63}(bds))$ was linked to a quartary olefinic carbon $(\delta_{C}^{138.0}(s))$



Methyl 3-hydroxy-4-methylpentanoate (II)

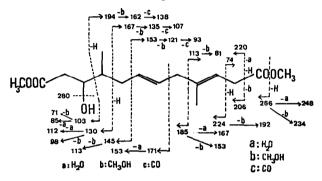
Methyl 4-methyl-3-pentenoate (III)

Spectral data of the isolated compound (Ib)

Spectral data of the isolated compound (1b)										
Carbon No.	¹³ C-nmr δppm	ⁱ H-Irradiation frequency(KHz)				(cm ⁻¹)	ms(m/e)			
1	173.5(s)				3500(bd)	2960(s)	93(100)			
2	37.8(t)	51.2677	2.40(2	(H,m)	2920(m)	1740(s)	107(38)	133(36)		
3	71.2(d)	51,4004	3.80(1		1735(s)	1438(m)	79(36)	74(34)		
4	38.2(d)	51,1928	1,64(1		1382 (m)	1367(m)	55(32)	147(32)		
5	35.3(t)	51,2353	2.08(2		1316(m)	1258(m)	81 (32)	130(29)		
Ğ	129.4(d)	51,5606	5.44(1		1200(m)	1165(m)	43(27)	193(22)		
2 3 4 5 6 7 8 9	129.9(d)	51,5606	5.44(1		1050(m)	1016(m)	41 (20)	135(17)		
8	42.5(t)	51,2899		H, bds)	997 (w)	972(m)	125(15)	121(14)		
9	138.0(s)		-		840(m)		103(13)	166(12)		
10	116.0(d)	51,5533	5,35(1	H.t.J=7.5H	z)		153(12)	206(12)		
11	33.4(t)	51,3244	2.99(2	H.d.J=7.5H	z)		152(10)	280(9)		
12	172.6(s)		•		•		146(8)	216(7)		
13	14.9(q)	51,1085	0.88(3	8H,d,J=7.0H	z)		266(6)	234(5)		
14	16.1(g)	51,1803	1.63(3	H,bds)			220(5)	248(4)		
15	51.6(a)	51,3905	3.74(3							
16	51.6(q)	51,3871	3.69(3	BH,s)						
		Spectral data	of meth	yl 3-hydro	xy-4-methy1~pe	ntanoate (11)			
Carbon	13 _{C-nmr}	1 _{H-nmr}			-1.					
No.	δppm	δppm		ir(c	:m ⁻¹)	ms	(m/e)			
1	173.5(s)	oppm		3500(bd)	2965(s)	103(100)	71 (93)	43(62)		
	38,2(t)	2,42(2H,m)		2880(m)	1740(s)			74 (30)		
2						bi(3b)	7.31281			
÷		3 80(1H.m)				61 (36)	73(28)	74(30)		
4	72.6(d)	3.80(1H,m) 1.66(1H sent'		1440(m)	1370(m)	61(36)	/3(28)	74(30)		
4	33.1(d)	1.66(1H,sept)		1440(m) 1280(m)	1370(m) 1173(m)	61(36)	73(28)	74(30)		
4 5 6	33.1(d) 17.7(q)	1.66(1H,sept) 0.95(3H,d,J=	7.0Hz)	1440(m) 1280(m) 1130(m)	1370(m) 1173(m) 1058(m)	61(36)	73(28)	/4(30)		
4 5 6 7	33.1(d) 17.7(q) 18.3(q)	1.66(1H,sept) 0.95(3H,d,J= 0.94(3H,d,J=	7.0Hz)	1440(m) 1280(m)	1370(m) 1173(m)	61(36)	73(28)	/4(30)		
2 3 4 5 6 7	33.1(d) 17.7(q)	1.66(1H,sept) 0.95(3H,d,J=	7.0Hz)	1440(m) 1280(m) 1130(m)	1370(m) 1173(m) 1058(m)	61(36)	73(28)	74(30)		
4 5 6 7	33.1(d) 17.7(q) 18.3(q)	1.66(1H,sept) 0.95(3H,d,J=7 0.94(3H,d,J=7 3.72(3H,s)	7.0Hz) 7.0Hz)	1440(m) 1280(m) 1130(m) 1015(m)	1 370 (m) 11 73 (m) 1058 (m) 998 (m)		73(28)	/4(30)		
	33.1(d) 17.7(q) 18.3(q) 51.7(q)	1.66(1H,sept) 0.95(3H,d,J=7 0.94(3H,d,J=7 3.72(3H,s)	7.0Hz) 7.0Hz)	1440(m) 1280(m) 1130(m) 1015(m) nyl 4-methy	1370(m) 1173(m) 1058(m) 998(m) -1-3-pentenoate	(111)				
	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s)	1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data	V.OHz) V.OHz) of meth	1440(m) 1280(m) 1130(m) 1015(m) nyl 4-methy 2960(s)	1370(m) 1173(m) 1058(m) 998(m) *1-3-pentenoate 2920(s)	(111) 69(100)	128(94)	68(31)		
	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s) 33.6(t)	1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data 2.97(2H,d,J=	7.0Hz) 7.0Hz) of meth 7.5Hz)	1440(m) 1280(m) 1130(m) 1015(m) nyl 4-methy 2960(s) 1737(s)	1370(m) 1173(m) 1058(m) 998(m) -1-3-pentenoate 2920(s) 1436(m)	(111)				
	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s) 33.6(t) 115.6(d)	1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data	7.0Hz) 7.0Hz) of meth 7.5Hz)	1440 (m) 1280 (m) 1130 (m) 1015 (m) my1 4-methy 2960 (s) 1737 (s) 1378 (m)	1370(m) 1173(m) 1058(m) 998(m) -1-3-pentenoate 2920(s) 1436(m) 1363(m)	(111) 69(100)				
	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s) 33.6(t) 115.6(d) 135.4(s)	1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data 2.97(2H,d,J= 5.30(1H,t,J=	7.0Hz) 7.0Hz) of meth 7.5Hz) 7.5Hz)	1440(m) 1280(m) 1130(m) 1015(m) 1015(m) 2960(s) 1737(s) 1378(m) 1315(m)	1370(m) 1173(m) 1058(m) 998(m) -1-3-pentenoate 2920(s) 1436(m) 1363(m) 1260(m)	(111) 69(100)				
	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s) 33.6(t) 115.6(d) 135.4(s) 25.6(q)	<pre>1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data 2.97(2H,d,J= 5.30(1H,t,J= 1.78(3H,d,J=)</pre>	7.0Hz) 7.0Hz) of meth 7.5Hz) 7.5Hz)	1440(m) 1280(m) 1130(m) 1015(m) 1015(m) 14-methy 2960(s) 1737(s) 1378(m) 1315(m) 1200(m)	1370(m) 1173(m) 1058(m) 998(m) *1-3-pentenoate 2920(s) 1436(m) 1363(m) 1260(m) 1162(m)	(111) 69(100)				
4 5 6 7 1 2 3 4 5 6 7	33.1(d) 17.7(q) 18.3(q) 51.7(q) 172.6(s) 33.6(t) 115.6(d) 135.4(s)	1.66(1H,sept 0.95(3H,d,J= 0.94(3H,d,J= 3.72(3H,s) Spectral data 2.97(2H,d,J= 5.30(1H,t,J=	7.0Hz) 7.0Hz) of meth 7.5Hz) 7.5Hz)	1440(m) 1280(m) 1130(m) 1015(m) 1015(m) 2960(s) 1737(s) 1378(m) 1315(m)	1370(m) 1173(m) 1058(m) 998(m) -1-3-pentenoate 2920(s) 1436(m) 1363(m) 1260(m)	(111) 69(100)				

Off-resonance ¹³C-nmr showed the presence of four methylenes which were assigned to be an α -carbonyl methylene (H₃COOC-<u>CH</u>₂-CH(OH)-; $\delta_C^{37.8(t)}$, $\delta_H^{2.40(m)}$, irradiation at 3.80 collapsed the m. at 2.40 to a d.), an α -carbonyl methylene (H₃COOC-<u>CH</u>₂-CH=C \leq ; $\delta_C^{33.4(t)}$, $\delta_H^{2.99(d)}$, irradiation at 5.35 simplified the d. at 2.99 to a s.), a methylene (-CH=CH-<u>CH</u>₂-C(CH₃)=CH-; $\delta_C^{42.5(t)}$, $\delta_H^{2.72(bds)}$, irradiation at 5.44 sharpened the bds. at 2.72) and a methylene (-<u>CH</u>₂-CH=CH-; $\delta_C^{35.3(t)}$, $\delta_H^{2.08(m)}$, irradiation at 5.44 simplified the m. at 2.08).

Consequently, the three partial structures, 2-carbomethoxy-1-hydroxy-ethyl moiety $(H_3COOC-CH_2-CH(OH)-)$, 7-carbomethoxy-5-methyl-2E,5-heptadienyl moiety $(-CH_2-CH=CH-CH_2-C(CH_3)=CH-CH_2-COOCH_3)$ and the methine carrying the methyl, assigned from above data, constructed the unique structure "dimethyl 3-hydroxy-4,9-dimethyl-6E,9-dodecadiene-dioate" for the structure of the new compound (Ib).



Examination of the precise mass determination was consistent with this structural feature. Ions at m/e $71(C_3H_3O_2)$, $103(CH(0H)CH_2COOCH_3)$ and a series of ions containing three oxygen atoms at m/e $224(M-CH_3COOCH_3)$, $185(M-C_4H_6COOCH_3)$, $171(M-C_5H_8COOCH_3)$, $145(M-C_7H_{10}COOCH_3)$ and $130(M-C_8H_{13}COOCH_3)$ were well explainable for the terminal 2-carbomethoxy-1-hydroxy-ethyl moiety. On the other hand, ions at m/e $113(C_4H_6COOCH_3)$, $153(C_7H_{10}COOCH_3)$, $167(C_8H_{12}COOCH_3)$ and $194(C_{10}H_{15}COOCH_3)$ were explainable to be due to the other terminal 3-carbomethoxy-1-methyl-1-propenyl moiety.

бррж 10-	H ^C J.		Rel	ative induce	d shift ratio	in LIS.	-1107
• 9-		Isolated compound (1b)			Methyl 3-hydroxy-4-methyl- pentanoate (II)		
8-		Carbo No.	ⁿ r _c	r _H	Carbon No.	r _c	r _H
7-	, C ² H ₂	ļ	3.66	7.73	1	3.74 5.88	7,76
6-	· C ^{6,7} H ₂	3	10.00	10,00	3	10.00	10.00
5-		4 5	0.37 0.99	4.33 2.88	4 5	0.13 1.00	4,15 2,67
	C ¹⁶ H2 C ⁴ H	6 7	0.94 0.42	1.44			
1		8	0.31	0.61			
3-	C ⁸ H ₂ -C ¹³ H ₃	10	0.26 0.42	0,76			
2-	<u>;</u> ;;C ¹⁴ H ₃	11 12	0.37 0.24	0.76			
1-		13 14	0.99	2.56 0.40	6	1.00	2.64
l	0,1 0.2 V 5 (6.1) / .	15	1.36	2.24	7	1.37	2,26
LIS	0.1 0.2 M, ratio Eu(fod) ₃ /comp. 5- ¹ H-nmar data of the isolated compound (Ib)	16	0.16	0.47			

Finally, the carbon framework of this compound was confirmed by a LIS-nmr experiment using the shift reagent "Eu(fod)₃". The metal was evidently co-ordinated with the hydroxy group, judged from the strongest lanthanide induced shift of the methine carrying the hydroxy group. Therefore, the strong LIS of the methylene ($\delta_{\rm C}^{37.8}$, relative induced shift ratio³) of the carbon=5.60; $\delta_{\rm H}^{2.40}$, $r_{\rm H}^{=7.73}$), the carbonyl ($\delta_{\rm C}^{173.5}$, $r_{\rm C}^{=3.66}$), the methoxy ($\delta_{\rm C}^{51.6}$, $r_{\rm C}^{=1.36}$; $\delta_{\rm H}^{3.74}$, $r_{\rm H}^{=2.24}$), the methyl ($\delta_{\rm C}^{14.9}$, $r_{\rm C}^{=0.99}$; $\delta_{\rm H}^{0.88}$, $r_{\rm H}^{=2.56}$), the methine ($\delta_{\rm C}^{38.2}$, $r_{\rm C}^{=0.37^4}$); $\delta_{\rm H}^{1.64}$, $r_{\rm H}^{=4.33}$) and the methylene ($\delta_{\rm C}^{35.3}$, $r_{\rm C}^{=0.99}$; $\delta_{\rm H}^{2.08}$, $r_{\rm H}^{=2.88}$), which were excellently agreeable with the LIS data of methyl 3-hydroxy-4-methyl-pentanoate (II), strongly supported the presence of the 4-carbomethoxy-3-hydroxy-2-methyl-butyl moiety in the structure.

The presence of the 3-carbomethoxy-1-methyl-1-propenyl moiety as the other terminal structure was well explainable by spectral data (ir, ¹³C- and ¹H-nmr) of methyl 4-methyl-3-pentenoate (III). The geometrical configuration of the trisubstituted double bond in the isolated compound was established to be trans, since the methyl (δ_{c} 16.1, δ_{H} 1.63 (bds)) was well consistent with the cis-methyl (δ_{c} 18.0, δ_{H} 1.67(bds)) rather than the trans-methyl (δ_{c} 25.6, δ_{H} 1.78(d,J=1.2Hz)) of methyl 4-methyl-3-pentenoate (III). ⁵) From the above data, 3 ξ -hydroxy-4 ξ ,9-dimethyl-6E,9E-dodecadienedioic acid was assigned for the structure of the isolated compound (Ia).

Judging from the symmetrical structure around the trans disubstituted double bond, this anomalous terpenoid acid is considered to be derived from the central part of the carotenoid chain such as phytoene.⁶

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- 2. ¹³C- and ¹H-nmr chemical shifts are expressed in ppm from TMS in CDCl₃. The assignments of ¹³C-nmr signals to ¹H-nmr signals were carried out by the selective decoupling ¹³C-nmr.
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- 4. The remarkable weak LIS of this methine carbon, which was also observed in the LIS of methyl 3-hydroxy-4-methyl-pentanoate (II), was considered to be due to the atomospheric shielding effect by the oxygen atom located between the metal and this methine carbon. The detail studies on these LIS data and on the stereo-chemistry of the isolated compound are in progress.
- 5. The assignment of two methyls was established by a LIS-nmr experiment which showed the differences between the cis-methyl (r_c =1.67; r_H =2.22) and the trans-methyl (r_c =1.39; r_H =1.11). The detail data containing ^C spin lattice relaxation time (T_1) of this compound is being published elsewhere.
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